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***MOLECULAR MECHANISMS INVOLVED IN  
METAL BIOACCUMULATION***

**Ph. D. THESIS ABSTRACT**

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**2014**

TABLE OF CONTENTS<sup>1</sup>

<b>Introduction</b>	1
<b>Part I. Theoretical considerations and literature data</b>	3
<b>I.1. <i>Saccharomyces cerevisiae</i> - eucharriot model used as biosorbent</b>	8
<b>I.1.1. The advantages of using <i>Saccharomyces cerevisiae</i> as metal biosorbent</b>	9
<b>I.1.2. Strains of <i>S. cerevisiae</i> cells used in bioremediation</b>	10
<b>I.1.3. Mechanisms used by <i>Saccharomyces cerevisiae</i> for heavy metal uptake</b>	11
<b>I.1.4. The effect of cell pretreatment on biosorbtion</b>	16
<b>I.1.5. External factors that influence the ability of <i>S. cerevisiae</i> cells to uptake metals</b>	17
<b>I.5.1.1. <i>pH</i></b>	17
<b>I.5.1.2. Initial metal ions concentration</b>	19
<b>I.5.1.3. Temperature</b>	19
<b>I.5.1.4. Contact time</b>	20
<b>I.5.1.5. Environmental conditions</b>	20
<b>I.5.1.6. Metallic ions characteristics</b>	20
<b>I.5.1.7. Growth media</b>	21
<b>I.1.6. Internal factors that influence the ability of <i>Saccharomyces cerevisiae</i> cells to uptake metals</b>	22
<b>I.1.6.1. Type of cells</b>	22
<b>I.1.6.2. Age of cells</b>	22
<b>I.2. Heavy metals</b>	23
<b>I.2.1. Heavy metals in the environment</b>	23
<b>I.2.2. Impact of heavy metals on the environment</b>	24
<b>I.2.3. Uptake of heavy metals from the environment</b>	24
<b>I.2.4. Toxicity of heavy metals</b>	25
<b>I.2.5. Heavy metals effect on the <i>S. cerevisiae</i> cells</b>	26
<b>I.2.5.1. Nickel</b>	26
<b>I.2.5.2. Cobalt</b>	27
<b>I.2.5.3. Manganese</b>	28
<b>I.2.5.3.1. Manganese ion transport in <i>Saccharomyces cerevisiae</i> cells</b>	29
<b>I.2.5.3.2. Phosphate ion transport in <i>Saccharomyces cerevisiae</i> cells</b>	34
<b>I.2.5.3.3. Phosphate-manganese connection: uptake of <math>Mn^{2+}</math> by <i>Saccharomyces cerevisiae</i> cells mediated by Pho84p</b>	36
<b>I.2.5.4. Cadmium</b>	38
<b>I.2.5.4.1. Transport and homeostasis of <math>Cd^{+2}</math> ions in <i>S. cerevisiae</i> cells</b>	38
<b>I.2.5.4.2. <math>Cd^{+2}</math> and antioxidant protection in <i>S. cerevisiae</i> cells</b>	43
<b>I.2.5.4.2.1. Effect on proteins</b>	43
<b>I.2.5.4.2.2. Effect on nucleic acids</b>	44
<b>I.2.5.5. Copper</b>	45
<b>I.3. Engineering of <i>S. cerevisiae</i> cells for increased heavy metal uptake</b>	48
<b>Part II. Original contributions</b>	50
<b>Chapter II.1. Accumulation of <math>Ni^{+2}</math> and <math>Co^{+2}</math> by mutant strains of <i>S. cerevisiae</i></b>	50
<b>II.1.A. General remarks</b>	51
<b>II.1.B. Results and discussions</b>	52
<b>II.1.B.1. Determination of toxic concentrations of <math>Me^{2+}</math> on the wild type strains of</b>	52

<sup>1</sup> The numbering of the pages is the one from the PhD thesis

	<i>S. cerevisiae</i>	
	II.1.B.2. Selection of mutant <i>S. cerevisiae</i> cells tolerant to high $\text{Me}^{2+}$ concentrations	53
	II.1.B.3. Characterization of $\text{Me}^{2+}$ tolerant strains	53
	II.1.B.4. Determination of $\text{Me}^{2+}$ accumulation by mutant cells	55
	II.1.B.5. Determination of $\text{Me}^{2+}$ intracellular distribution in <i>nir4</i> , <i>nir5</i> and <i>cor5</i> mutants	56
	II.1.B.6. The <i>nir</i> and <i>cor</i> mutants decrease $\text{Me}^{2+}$ concentration in environment	57
	II.1.B.7. Conclusions	58
II.1.C.	Experimental part	59
	II.1.C.1. Strains and growth conditions	59
	II.1.C.2. Evaluation of cell growth in the liquid media supplemented with metallic ions	60
	II.1.C.3. Cell growth in spot experiments	60
	II.1.C.4. Mutagenesis and selection of $\text{Me}^{2+}$ resistant mutants cells	60
	II.1.C.5. Tetrad dissection	60
	II.1.C.6. Differential extraction of $\text{Me}^{2+}$ from cytosol and vacuoles	61
	II.1.C.7. Determination of $\text{Me}^{2+}$ concentration accumulated in cells	61
	II.1.C.8. Determination of cellular protein concentration	62
Chapter II.2.	Use of “kamikaze” cells for uptake of $\text{Mn}^{2+}$ , $\text{Co}^{2+}$ , $\text{Cu}^{2+}$ and $\text{Cd}^{2+}$ from synthetic effluents	64
II.2.A.	General remarks	65
II.2.B.	Results and discussion	66
	II.2.B.1. Cellular response of <i>pmr1Δ</i> cells in presence of heavy metals	66
	II.2.B.2. Influence of $\text{Mn}^{2+}$ , $\text{Cu}^{2+}$ and $\text{Co}^{2+}$ on <i>pmr1Δ</i>	67
	II.2.B.3. Influence of $\text{Cd}^{2+}$ on <i>pmr1Δ</i> cells	69
	II.2.B.4. Ability of <i>Δpmr1</i> cells to take up metal ions from synthetic effluents	70
	II.2.B.5. Conclusions	76
II.2.C.	Experimental part	78
	II.2.C.1. Strains, media and growth conditions	78
	II.2.C.2. The accumulation of heavy metals by cells	78
	II.2.C.3. Assay of metallic ions accumulated in cells	79
	II.2.C.4. Assay of cellular proteins	79
	II.2.C.5. Bioremediation of synthetic effluents using cells in a batch system	79
Chapter II.3.	Effect of <i>PHO84</i> overexpression on the metal accumulation	81
II.3.A.	General remarks	83
II.3.B.	Results and discussions	84
	II.3.B.1. <i>S. cerevisiae</i> overexpressing <i>PHO84</i> gene	84
	II.3.B.2. Effect of <i>PHO84</i> overexpression on the cell growth	84
	II.3.B.3. Overexpression of <i>PHO84</i> gene under <i>GALI</i> promoter control triggers increase of heavy metals accumulation in <i>S. cerevisiae</i> cells	85
	II.3.B.4. Overexpression of <i>PHO84</i> triggers UPR pathway	89
	II.3.B.5. Localization of Pho84p in <i>PHO84</i> overexpressing cells	90
	II.3.B.6. Overexpression of <i>PHO84</i> in <i>ire1Δ</i> cells	91
	II.3.B.7. Phosphate accumulation in <i>PHO84</i> -overexpressing cells	93
	II.3.B.8. Overexpression of <i>PHO84</i> in <i>pmr1Δ</i>	94
	II.3.B.9. Conclusions	97
II.3.C.	Experimental part	99
	II.3.C.1. Strains, media and growth conditions	99
	II.3.C.2. Plasmids	100
	II.3.C.3. Genomic DNA isolation from <i>S. cerevisiae</i> cells	101
	II.3.C.4. Amplification of <i>PHO84</i> gene by PCR	101
	II.3.C.5. Electrophoresis of PCR products	102
	II.3.C.6. Purification of PCR products	103

II.3.C.7. Cloning PCR products	103
II.3.C.8. Transformation of <i>Escherichia coli</i> cells	103
II.3.C.9. Isolation of plasmids from ampicilin rezistent colonies	103
II.3.C.10. Purification of DNA sequences from agarose gels	104
II.3.C.11. Transformation of <i>S. cerevisiae</i> cells	104
II.3.C.12. Obtaining of pPHO84-pGREG505/506 plasmids	105
II.3.C.13. Obtaining pPHO84-pGREG600 plasmid	106
II.3.C.14. Growth assessments of <i>S. cerevisiae</i> on selective media	107
II.3.C.15. Accumulation of metals by cells	107
II.3.C.16. Localization of Pho84p-GFP construct	108
II.3.C.17. $\beta$ -galactosidase assay	108
II.3.C.18. Phosphate ions assay	109
II.3.C.19. Reproducibility of results	110
II.3.C.20. Statistic analysis	111
<b>Chapter II.4. Calcium signaling mediates the response to cadmium toxicity in <i>S. cerevisiae</i> cells</b>	<b>113</b>
II.4.A. General remarks	115
II.4.B. Results and discussion	116
II.4.B.1. Yeast cells exposed to high concentration of $\text{Cd}^{2+}$ respond through a transient increase of cytosolic $\text{Ca}^{2+}$	116
II.4.B.2. $\text{Ca}^{2+}$ mediated response to $\text{Cd}^{2+}$ ions depends of $\text{Ca}^{2+}$ exogen	117
II.4.B.3. $[\text{Ca}^{2+}]_{\text{cyt}}$ pulses induce by $\text{Cd}^{2+}$ occurs faster $\text{Cd}^{2+}$ uptake	119
II.4.B.4. Mutants with defects in $\text{Ca}^{2+}$ homeostasis gene have different tolerance to $\text{Cd}^{2+}$	122
II.4.B.5. $[\text{Ca}^{2+}]_{\text{cyt}}$ regulates $\text{Cd}^{2+}$ response	124
II.4.B.6. The $\text{Cd}^{2+}$ tolerance/sensitivity of mutants with defects in oxidative stress response correlate with $[\text{Ca}^{2+}]_{\text{cyt}}$ pulse	125
II.4.B.7. Conclusions	128
II.4.C. Experimental part	129
II.4.C.1. Strains, media, growth conditions and plasmids	129
II.4.C.2. <i>In vivo</i> monitoring of calcium pulse induced by metallic stress	129
II.4.C.3. Cell growth assessment on solid media	130
II.4.C.4. The uptake of $\text{Cd}^{2+}$ by living cells	130
II.4.C.5. Reproducibility of results	131
II.4.C.6. Statistic analysis	131
<b>Chapter II.5. <i>Vaccinium corymbosum</i> L. (blueberry) extracts exhibit protective action against cadmium toxicity in <i>S. cerevisiae</i> cells</b>	<b>133</b>
II.5.A. General remarks	135
II.5.B. Results and discussions	136
II.5.B.1. <i>Vaccinium corymbosum</i> extract protect cells against $\text{Cd}^{2+}$ toxicity	136
II.5.B.2. The <i>yap1<math>\Delta</math></i> hypersensitivity to $\text{Cd}^{2+}$ is alleviated by <i>V. corymbosum</i> extract in a dose dependent manner	138
II.5.B.3. Cyanidin effect against $\text{Cd}^{2+}$ toxicity in the yeast cells	140
II.5.B.4. <i>V. corymbosum</i> extract and cyanidin protect yeast cells against $\text{Cd}^{2+}$ toxicity	142
II.5.B.5. Conclusions	144
II.5.C. Experimental part	144
II.5.C.1. Preparation of <i>V. corymbosum</i> extracts	144
II.5.C.2. Total phenols assay	140
II.5.C.3. Total antocyanidin assay	145
II.5.C.4. Strains and growth conditions	145
II.5.C.5. Growth assessment	145
II.5.C.5.1. Growth in liquid media	146
II.5.C.5.2. Cells growth spot assay	146

II.5.C.6. Halo assay	146
II.5.C.7. Cell viability	146
II.5.C.8. Metal accumulation in living cells	146
II.5.C.9. <i>V. corymbosum</i> extract uptake in <i>S. cerevisiae</i> cells	147
II.5.C.10. Reproducibility of results	147
<b>Annex 1</b>	151
A.1. Obtaining a hyperaccumulator <i>Saccharomyces cerevisiae</i> strain that overexpress the <i>PHO84</i> gene	151
A.2. The pGREG505 plasmid map	160
A.3. The pGREG506 plasmid map	161
A.4. The pGREG600 plasmid map	162
A.5. Obtaining a <i>pmc1Δ vcx1Δ</i> double null mutant	
<b>Annex 2</b>	163
Solutions and growth media	163
<b>Annex 3</b>	168
Strains	168
<b>Bibliography</b>	171
<b>General conclusions</b>	180

2

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<sup>2</sup> The numbering of tables, figures and references are as in the Ph. D. thesis.

The thesis consists of two main parts: **Theoretical consideration**, which describes the data from the literature on research topics addressed and **Original contributions**, presenting the personal contributions.

The first subchapter presents the general aspects related to the eukaryotic model *Saccharomyces cerevisiae* used as a potential biosorbent. In this part the external and internal factors that affect the ability of *Saccharomyces cerevisiae* to uptake metals from contaminated effluents, particularly aquatic environment are presented.

In the second subchapter I have discussed upon data and information about heavy metals, highlighting the sources of heavy metals in the environment, the impact of heavy metals upon the environment, removal of metals from the environment, heavy metal toxicity and effect of heavy metals on cells of *S. cerevisiae*.

In the third subchapter, ways to engineer *S. cerevisiae* cells to increase capacity to accumulate heavy metals are presented.

The main objective of this thesis was to obtain strains of *Saccharomyces cerevisiae* capable to accumulate significant amounts of potentially polluting metal ions and to develop cellular systems involved in bioremediation of waste water contaminated with heavy metals. This was achieved by studying the molecular mechanisms involved in the bioaccumulation of metal ions.

The work presented in this thesis was focused on five directions of research.

Metal remediation through common physico-chemical techniques is expensive and unsuitable in case of voluminous effluents containing complex organic matter and low metal contamination. Alternative biotechnological approaches received great deal of attention in the recent years. Engineering cell lines that would hyperaccumulate heavy metals can be an invaluable tool in removing such ions from aqueous environments.

In the first direction of research, the production of mutant strains of *S. cerevisiae* resistant to high concentrations of heavy metals that would accumulate heavy metals in a non-toxic manner was attempted. As a result of the studies we obtained 2 strains tolerant to high concentration of  $\text{Ni}^{2+}$  and one strain tolerant to high concentrations of  $\text{Co}^{2+}$ . These strains had the ability to accumulate metals in vacuoles. An important aspect was the fact that the strains were able to reduce the amount of  $\text{Me}^{2+}$  from media in a single cycle of growth, which made them good candidates for use in bioremediation processes.

The second direction of research focused on the possibility of using “kamikaze” strains with high potential for bioaccumulation of heavy metals, but killed in the process of bioremediation. The most interesting was the strain defective in the ATP-ase pump Pmr1p (responsible for detoxification of metals by excluding them by cellular secretory pathway). It was shown that null-mutant strain *pmr1Δ* had an increased capacity to remove  $Mn^{2+}$ ,  $Cu^{2+}$ ,  $Co^{2+}$  or  $Cd^{2+}$  from synthetic effluents due to the ability to hyperaccumulate these cations. Due to increased metal accumulation, the mutant strains was more efficient than the wild type in removing heavy metals containing 1-2 mM cations, with a selectivity  $Mn^{2+} > Co^{2+} > Cu^{2+}$  and also in removing  $Mn^{2+}$  and  $Cd^{2+}$  from synthetic effluents containing 20-50  $\mu M$  cations, with a selectivity  $Mn^{2+} > Cd^{2+}$ .

It was also found that the *pmr1Δ* cells had a tendency to accumulate large amounts of metal ions as compared to cells that normally express the *PMR1* gene.

The third direction sought to obtain strains that hyperaccumulate heavy metals, not by deletion of genes involved in cellular detoxification, but rather by overexpression of genes encoding transporters for metal ions. In addition, we studied the effects of the gene overexpression on the ability of cells to bioaccumulate metal ions. Pho84p, the protein responsible for the high-affinity uptake and transport of inorganic phosphate across the plasma membrane, is also involved in the low-affinity uptake of heavy metals in the *Saccharomyces cerevisiae* cells.

This part of the thesis demonstrated that under metal ions excess, yeast cells overexpressing *PHO84* gene acquire an increased capacity to bioaccumulate  $Mn^{2+}$ ,  $Cu^{2+}$  or  $Co^{2+}$  and in some genetic background cells becomes hyperaccumulators. As *PHO84* overexpression triggered the Ire1p-dependent unfolded protein response, abundant plasma membrane Pho84p could be achieved only in *ire1Δ* cells (lacking the gene that encodes a transmembrane kinase which transmit the signal about unfolded proteins to RE). Under environmental surplus, *PHO84* overexpression augmented the metal accumulation by the wild type, accumulation that was exacerbated by the *IRE1* deletion. The *pmr1Δ* cells (lacking the gene that encodes the P-type ATPase ion pump that transports  $Ca^{2+}$  and  $Mn^{2+}$  into the Golgi), hyperaccumulated  $Mn^{2+}$  even from normal medium when overexpressing *PHO84*, a phenotype which is rather restricted to metal-hyperaccumulating plants.

For a better understanding of the mechanisms involved in cell survival and adaptation to stress caused by heavy metals, in the fourth direction of research the involvement of  $\text{Ca}^{2+}$  in signaling the cell exposure to high concentrations of metals was studied. Using the *S. cerevisiae* cells which expressed a transgenic  $\text{Ca}^{2+}$ -sensitive photoprotein it was found that among the metals tested, only  $\text{Cd}^{2+}$  surplus was signaled by calcium.

The yeast cells responded through a sharp increase in cytosolic  $\text{Ca}^{2+}$  when exposed to  $\text{Cd}^{2+}$ , and to a lesser extent to  $\text{Cu}^{2+}$ , but not to  $\text{Mn}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Zn}^{2+}$ , or  $\text{Hg}^{2+}$ . The response to high  $\text{Cd}^{2+}$  depended mainly on external  $\text{Ca}^{2+}$  (transported through the Cch1p/Mid1p channel), but also on vacuolar  $\text{Ca}^{2+}$  (released into the cytosol through the Yvc1p channel). The adaptation to high  $\text{Cd}^{2+}$  was influenced by perturbations in  $\text{Ca}^{2+}$  homeostasis. The data obtained in this part of the thesis indicate that the presence of high concentrations of  $\text{Cd}^{2+}$  in the environment is signaled through immediate and sudden pulses of cytosolic  $\text{Ca}^{2+}$ . Apparently, sudden and sharp pulses of  $\text{Ca}^{2+}$  allow the adaptation to high  $\text{Cd}^{2+}$ , while the absence of  $[\text{Ca}^{2+}]_{\text{cyt}}$  signaling or broad pulses and lingering  $[\text{Ca}^{2+}]_{\text{cyt}}$  are responsible for  $\text{Cd}^{2+}$  hypersensitivity.

Since some cell strains that were shown to be hyperaccumulators of heavy metals die due to the toxicity of those metals, in a fifth direction of research, we studied the using of plant antioxidants to protect cells from the stress induced by heavy metals.

*Vaccinium corymbosum* L. are a rich source of antioxidants and their consumption is believed to contribute to food-related protection against oxidative stress. Four varieties of blueberries were used in the study, and it was found that the extracts with high content of total anthocyanidins exhibited significant protective effect against the toxicity of cadmium and  $\text{H}_2\text{O}_2$ . Both the blueberry extracts and pure cyanidin exhibited protective effects against cadmium in a dose-dependent manner, but without significantly interfering with the cadmium accumulation by the yeast cells. Thus, it was proved that the extract of *Vaccinium corymbosum* berries has a protective effect against *S. cerevisiae* cells exposed to  $\text{Cd}^{2+}$  ions, one of the most toxic metals studied. The results imply that the blueberry extracts might be a potentially valuable food supplement for individuals exposed to high cadmium.



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